

Magnetic Interaction between Adsorbed O₂ Molecules in [Cu₂(bza)₄(pyz)]_n

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Recently, [Cu₂(bza)₄(pyz)]_n (bza=benzoate, pyz=pyrazine), which can generate stable O₂-inclusion crystals at low temperature using forcible adsorption conditions, has been extensively investigated.[1,2] In this system, interesting magnetization was found, and the trimer induced by magnetic interactions between adsorbed O₂ molecules in the nano-porous cavity was expected. In the present work, we carried out neutron scattering measurements on [Cu₂(bza)₄(pyz)]_n to clarify the magnetic interaction between adsorbed O₂ molecules. Neutron measurements were carried out using the cold neutron multi-chopper spectrometer Let installed at ISIS. After measurements on the [Cu₂(bza)₄(pyz)]_n polycrystalline sample sealed with high pressure oxygen gas (60 atm at room temperature), the measurements on the [Cu₂(bza)₄(pyz)]_n having no O₂ were also performed.

$S(Q, E)$ measured at 1.8 K for [Cu₂(bza)₄(pyz)]_n polycrystalline sample sealed with 60 atm oxygen gas indicates the dispersion-less excitations at $E \sim 0.4$ meV and 4 meV. Since these excitations were not observed for the [Cu₂(bza)₄(pyz)]_n having no O₂, these are originated from the magnetic moments of O₂. The constant-E scans show that both excitations have the peaks at $Q \sim 1.1 \text{ \AA}^{-1}$.

We study the observed magnetic excitations using the trimer model. The spin Hamiltonian is

$$\mathcal{H}_{trimer} = J_1 \mathbf{S}_1 \cdot \mathbf{S}_2 + J_2 \mathbf{S}_2 \cdot \mathbf{S}_3 + D \sum_l (S_l^z)^2. \quad (1)$$

As the result of the fitting, we obtained the exchange interactions $J_1=3.98$ meV, $J_2=0.79$ meV, and the uniaxial anisotropy $D=0.44$ meV. The magnetic excitation can be rea-

sonably explained by the trimer model. Although there is the large difference between the values of J_1 and J_2 , the similar difference has been reported in the theory about the Rh-system having the randomness of S₂ site.[2] On the other hand, the temperature dependence of observed excitations is no simple. At this moment, we try to make the analyses of the T -dependence in order to clarify the magnetic correlation of trimer.

[1] S. Takamizawa, E. Nakata, and T. Akutsuka, *Andew. Chem., Int. Ed.* 45 (2006) 2216.

[2] S. Takamizawa et al., *J. AM. CHEM. SOC.* 130 (2008) 17883.